CHROM. 8434

Note

Steroids and related studies

XXXII*. Thin-layer chromatography of some lactam, tetrazole, basic and quaternary azasteroids

HARKISHAN SINGH, DHARAM PAUL, RAVINDER K. MALHOTRA and TILAK RAJ BHARDWAJ

Department of Pharmaceutical Sciences, Panjab University, Chandigarh 160014 (India)
(Received December 30th, 1974 and February 26th, 1975; revised manuscript received May 1st, 1975)

Our laboratory is active in the synthesis of azasteroids, and recently several lactams^{2,3}, tetrazoles^{1,4}, and basic and quaternary compounds^{2,3} have been prepared. The purity of these compounds was ascertained by TLC, and in this paper we report the results of the chromatographic investigations; other communications on the TLC patterns of steroidal lactams have been published^{5,6}.

EXPERIMENTAL

Reagents

All the reagents were of analytical grade and were used as such.

Azasteroids

The lactams (Table I), tetrazoles (Table II) and basic (Table III) and quaternary (Table IV) azasteroids used were synthesized in this laboratory.

Method

The plates ($200 \times 200 \times 3.5$ mm) were coated with silica gel G (Merck) to a thickness of 0.25 mm and were activated at 110° for 1 h, then stored in a drying cabinet over calcium chloride.

Solutions of the azasteroids were prepared in methanol or chloroform and were applied to give $50-100 \mu g$ of the material as a spot on the plate.

The nine solvent systems used were chloroform, methanol, chloroform—methanol, chloroform—ethyl acetate, ethyl acetate—n-hexane, methanol—ammonia, benzene—methanol—ethyl acetate, ethanol—chloroform—ethyl acetate—water—hydro-chloric acid, and n-butanol—acetic acid—water.

The following three systems proved to be of particular interest for steroidal lactams and tetrazoles:

(1) chloroform-methanol (19:1).

^{*} For the previous paper in this series, see ref. 1.

- (2) benzene-methanol-ethyl acetate (17:2:1), and
- (3) ethyl acetate-n-hexane (3:2).

The systems found to be appropriate for basic and quaternary azasteroids were:

- (4) methanol-30% (w/w) ammonia solution (9:1),
- (5) ethanol-chloroform-ethyl acetate-water-concentrated hydrochloric acid (12:12:12:1), and
 - (6) *n*-butanol-glacial acetic acid-water (5:3:2).

The chromatograms were run at room temperature ($\simeq 35^{\circ}$), and spots were detected with use of ceric sulphate solution (2 g in 100 ml of 10% (v/v) sulphuric acid), followed by heating at 150°, and by exposure to iodine vapour; permanent black spots were obtained with the ceric sulphate spray reagent. The development of spots with iodine vapour took only 2 to 4 min, and the spots were brown.

RESULTS

Tables I, II, III and IV list the R_F values of lactams, tetrazoles and basic and quaternary azasteroids, respectively. For the lactams, solvent systems 1 and 2 gave consistent results. For the tetrazoles, solvent system 3 was also found to be of interest. Solvent systems 4, 5 and 6 gave consistent results for basic azasteroids, and systems 5 and 6 were useful for quaternary compounds.

TABLE I
THIN-LAYER CHROMATOGRAPHY OF STEROIDAL LACTAMS IN SOLVENT SYSTEMS
1 AND 2

Compound	R _F value in solvent	
	1	2
4-Aza-5α-cholestan-3-one	0.56	0.43
17-Oxo-17a-aza-D-homoandrost-5-en-3β-yl acetate	0.43	0.34
3 <i>β</i> -Hydroxy-17a-aza-D-homoandrost-5-en-17-one	0.22	0.19
17a-Aza-D-homoandrost-4-ene-3,17-dione	0.32	0.26
3-Hydroxyimino-17a-aza-D-homoandrost-4-en-17-one	0.20	0.26
1.17a-Diaza-D-homo-5a-androstane-3.17-dione	0.06	0.08
4-Hydroxy-4,17a-diaza-D-homo-5a-androstane-3,17-dione	0.10	0.10
4-Benzyl-4,17a-diaza-D-homoandrost-5-ene-3,17-dione	0.31	0.29
3,5-Seco-5,17-dioxo-17a-aza-D-homo-4-norandrostan-3-oic acid		0.03
3,5-Seco-5-hydroxyimino-17-oxo-17a-aza-D-homo-4-norandrostan-3-oic acid		0.03

The results suggest that solvent system 2 is useful for the TLC of steroidal lactams; this observation is in agreement with the work of Doorenbos and Sharma⁵. For steroidal tetrazoles, systems 1 and 3 can also be used with good results. System 1 may be particularly useful for detecting lactams and tetrazoles in mixtures resulting from the Schmidt reaction with steroidal ketones.

The quaternary compounds generally move slower than the related tertiary amines in systems 5 and 6, so that quaternisation experiments can be monitored, as was done by Bamford *et al.*⁷, who used system 5.

272 NOTES

TABLE II
THIN-LAYER CHROMATOGRAPHY OF STEROIDAL TETRAZOLES IN SOLVENT SYSTEMS 1, 2 AND 3

Compound	R _F value in solvent		
	1	2	3
3-Aza-A-homo-4a-cholesteno[3,4-d]tetrazole	0.73	0.73	0.86
(25R)-3-Aza-A-homo-4a-spirosteno[3,4-d]tetrazole	0.72	0.72	0.75
3-Aza-A-homo-4a, 16-pregnadieno[3,4-d]tetrazol-20-one	0.66	0.55	0.47
3-Aza-A-homo-4a,16-pregnadieno[3,4-d]tetrazol-20-one oxime	0.52	0.53	0.68
17β-Acetamido-3-aza-A-homo-4a-androsteno[3,4-d]tetrazole	0.37	0.25	0.02
17β -(5-Methyltetrazol-1-yl)-3-aza-A-homo-4a-androsteno[3,4-d]te-			
trazole	0.42	0.32	0.05
3-Aza-A-homo-4a-androsteno[3,4-d]tetrazol-17-one	0.59	0.51	0.29
3-Aza-A-homo-4a-androsteno[3,4-d]tetrazol-17 β -ol	0.38	0.37	0.27
3-Aza-A-homo-4a-androsteno[3,4-d]tetrazol-17 β -yl acetate	0.73	0.65	0.61
3-Aza-A-homo-4a-androsteno[3,4-d]tetrazol-17-one oxime	0.40	0,37	0.24
3.17a-Diaza-A.D-bishomo-4a-androsteno[3,4-d]tetrazol-17-one		0.13	
3,17a-Diaza-A,D-bishomo-4a-androsteno[3,4-d][17a,17-d]bistetrazole	0.65	0.43	0.28
13,17-Seco-13a-azido-A-homo-4a-androsteno[3,4-d]tetrazole 17-nitrile	0.66	0.43	0.26
3-Methoxy-17a-aza-D-homo-oestra-1,3,5(10)-trieno[17a,17-d]tetrazole	0.67	0.41	0.28
7a-Aza-B-homocholest-5-eno[7a,7-d]tetrazol-3β-yl acetate	0.75	0.69	0.77
7a-Aza-B-homocholest-4-eno[7a,7-d]tetrazol-3 β -yl acetate	0.77	0.70	0.71
7a-Aza-B-homocholest-4-eno[7a,7-d]tetrazol-3β-ol	0.29	0.26	_
7a-Aza-B-homocholest-5-eno[7a,7-d]tetrazol-3 β -ol	0.38	0.32	0.17
7a-Aza-B-homocholest-4-eno[7a,7-d]tetrazol-3-one	0.71	0.47	0.40

TABLE III
THIN-LAYER CHROMATOGRAPHY OF BASIC AZASTEROIDS IN SOLVENT SYSTEMS
4, 5 AND 6

Compound	R _F value in solvent		
	4	5	6
4-Aza-5α-cholestane	0.42	0.49	0.67
4-Methyl-4-aza-5α-cholestane	_	0.38	0.40
4-Aza- 5α -androstan- 17β -ol	0.41	0.54	0.57
4-Methyl-4-aza-5α-androstan-17β-ol	0.66	0.40	0.46
4-Methyl-4-aza-5 α -androstan-17 β -ylacetate	0.68	0.46	0.47
3β-Hydroxy-17a-aza-D-homoandrost-5-enc	0.25	0.49	0.55
17a-Methyl-17a-aza-D-homoandrost-5-en-3β-ol	0.45	0.40	0.42
17a-Methyl-17a-aza-D-homoandrost-4-en-3-one	0.45	0.32	0.35
17a-Methyl-17a-aza-D-homoandrost-5-en-3β-yl acetate	0.47	0.49	0.47
17a-Methyl-17a-aza-D-homo-4a-androsteno[3,4-d]tetrazole	0.42	0.25	0.25
4.17a-Diaza-D-homo-5α-androstane	0.08	0.13	0.17
4,17a-Dimethyl-4,17a-diaza-D-homo-5α-androstane	0.32	0.03	0.08
38-Pyrrolidino-17a-aza-D-homoandrost-5-ene	0.19	0.19	0.23
3β-Pyrrolidino-17a-methyl-17a-aza-D-homoandrost-5-ene	0.34	0.10	0.13
ββ-Pyrrolidino-17a-aza-D-homoandrost-5-en-17-one	0.62	0.27	0.36
7β -Amino-4-aza-5 α -androstane	0.28	0.18	0.30
4-Methyl-17β-dimethylamino-4-aza-5α-androstane	0.61	0.06	0.13

TABLE IV
THIN-LAYER CHROMATOGRAPHY OF QUATERNARY AZASTEROIDS IN SOLVENT SYSTEMS 5 AND 6

Compound	R_F val	R _F value in solvent	
	5	6	
4-Methyl-4-aza-5α-cholestane methiodide	0.47	0,56	
4-Methyl-4-aza-5 α -androstan-17 β -ol methiodide	0.16	0.26	
4-Methyl-4-aza-5 α -androstan-17 β -yl acetate methiodide	0.29	0.28	
17a-Methyl-17a-aza-D-homoandrost-5-en-3β-ol methiodide	0.17	0.24	
17a-Methyl-17a-aza-D-homoandrost-4-en-3-one methiodide	0.11	0.16	
4,17a-Dimethyl-4,17a-diaza-D-homo-5α-androstane dimethiodide		0.02	
17a-Methyl-3β-pyrrolidino-17a-aza-D-homoandrost-5-ene dimethiodide		0.05	
3β-Pyrrolidino-17a-aza-D-homoandrost-5-en-17-one methiodide	0.10	0.26	
4-Methyl-17 β -dimethylamino-4-aza-5 α -androstane dimethiodide		0.08	

ACKNOWLEDGEMENT

We thank the Council of Scientific and Industrial Research, New Delhi, for financial support.

REFERENCES

- 1 H. Singh, R. K. Malhotra and V. V. Parashar, Indian J. Chem., in press.
- 2 H. Singh, D. Paul and V. V. Parashar, J. Chem. Soc., Perkin Trans. I, (1973) 1204.
- 3 H. Singh and D. Paul, J. Chem. Soc., Perkin Trans. 1, (1974) 1475.
- 4 H. Singh, R. K. Malhotra and N. K. Luhadiya, J. Chem. Soc., Perkin Trans. I, (1974) 1480.
- 5 N. J. Doorenbos and R. K. Sharma, J. Chromatogr., 29 (1967) 393.
- 6 B. Matkovics and Zs. Tegyey, Magy. Kém. Foly., 74 (1968) 516.
- 7 D. G. Bamford, D. F. Biggs, M. Davis, and E. W. Parnell, J. Pharm. Pharmacol., 23 (1971) 595.